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Dissolved iodine in rainfall, cloud, stream and groundwater in the Plynlimon area of mid-Wales

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Abstract

Iodine concentrations in rainfall, cloud water, throughfall, stemflow, stream water and groundwater are described for the upper River Severn Plynlimon catchments in mid-Wales. Iodine concentrations average $1.55 \mu\text{g l}^{-1}$ in rainfall while the corresponding averages of iodine concentrations in cloud water, throughfall and stemflow iodine concentrations are three to eight times higher. For the atmospheric inputs, there are both marine and anthropogenic sources while, for stemflow and throughfall, humic substances may be a dominant component. The average concentrations in the streams and groundwaters vary between 1.00 to $4.22 \mu\text{g l}^{-1}$ and 0.57 to $8.51 \mu\text{g l}^{-1}$, respectively. The higher averages are for waters associated with gley soils that are high in dissolved organic carbon (DOC). For the streams draining forested catchments, iodine concentrations have increased over time, probably due to wetting up of the soils because of reduction in transpiration as the trees are maturing and felling activities. The increases vary between 0.025 ± 0.003 and $0.044 \pm 0.010 \mu\text{g l}^{-1}$ per year and are associated partly with an increase in DOC concentration: adjusted for the effect of DOC, the increases range between 0.013 ± 0.003 and $0.023 \pm 0.009 \mu\text{g l}^{-1}$ per year. For the moorland areas, no such trend is observed but the value is lower. For most of the Plynlimon catchments, a net uptake of iodine of around 40% is lower for streams draining gley soils with even a net export for the smaller streams. Within the streams, iodine concentrations are highly correlated with DOC and there is an annual cycle of iodine and DOC concentration change, with maxima between July and November. For the groundwaters, a loss of iodine and DOC relative to the streams probably reflects within catchment losses in the lower inorganic soils and bedrock areas. However, in one case, groundwater iodine concentrations were relatively high and this coincided with high DOC concentrations.

Keywords: dissolved organic carbon, Hafren, Hore, iodide, iodine, cloud, mist, rain, river, stemflow, Tanllwyth, throughfall, catchment wetness

Introduction

Iodine is an important micro-nutrient in the environment and deficiencies are associated with health risks for both humans and wildlife (Jooste *et al.*, 1992; Edmunds and Smedley, 1996; Phaneuf *et al.*, 1999; Milewski and Diamond, 2000; Sack *et al.*, 2000). Further, there are issues of radioactive iodine contamination from nuclear weapon testing, nuclear accidents and emissions from the nuclear power industry. A global distribution of radioactive iodine occurs because of the volatility of many of the iodine compounds involved and because particulate iodine in the atmosphere is frequently associated with very small aerosols (Radlinger and Heumann, 2000). Iodine is a relatively rare element in the earth's crust averaging around 0.46 ppm and is usually found in minerals as the anion iodate (Greenwood and Ernschaw, 2005). Iodine concentration in seawater

averages around $50 \mu\text{g l}^{-1}$ and the concentrations in rainfall and river water are usually an order of magnitude lower (Moran *et al.*, 2002). However, much higher concentrations can occur in brines associated with oil fields where concentrations can average 30 000 to 40 000 $\mu\text{g l}^{-1}$ (Greenwood and Ernschaw, 2005).

A major proportion of iodine in many surface fresh waters is generally thought to have been derived from the sea via the atmosphere (Moran *et al.*, 2002); the transfer is complex, involving both inorganic and organic components and the processes include sea spray and biogenic production of organo-iodine (Radlinger and Heumann, 2000). Iodine is concentrated in soils, especially organic rich ones, but can be released back to solution under reducing conditions (Muramatsu and Yoshida, 1999; Yuita and Kihou, 2005). Iodine exists in a number of oxidation states: -1 , 0 , 1 , 3 , 5 and 7 . However, it most often occurs in natural waters

inorganically in the -1 (iodide) and +5 (iodate) oxidation states which are the most stable thermodynamically (Jones, 1981; Jones and Truesdale, 1984; Truesdale and Jones, 1996; Moran *et al.*, 2002). Iodine can also be present in significant concentrations as organo-iodine in waters and fixed onto humic substances (Warwick *et al.*, 1993; Radlinger and Heumann, 2000; Andersen *et al.*, 2002).

In this paper, the distribution of iodine in rainfall, cloud water (mist), stream water and groundwater is presented for the Plynlimon catchments in mid-Wales, to describe iodine water quality for a classic upland typology. The work develops from an earlier study of halogens in the area (Neal *et al.*, 1990) by showing longer term (decadal) patterns of change that link to increasing dissolved organic carbon concentrations (Neal *et al.*, 2005). It parallels the study of bromine concentrations in the Plynlimon area (Neal *et al.*, 2007) and earlier studies of iodine species in rainfall and surface waters at Plynlimon and other parts of the UK (Jones, 1981; Jones and Truesdale, 1984; Truesdale and Jones, 1996). The work provides new information on iodine in groundwater at Plynlimon for a shallow fracture flow dominated aquifer in a hard rock terrain.

Study area and monitoring programme

The study is based in the headwater areas of the River Severn in mid-Wales in the Plynlimon experimental catchments. Iodine concentrations in these catchments have been determined in rainfall, cloud water (mist), stemflow, throughfall, stream water and groundwater as part of ongoing long-term (10 to 25 years) baseline monitoring as well as one to five year more focussed monitoring programmes (Neal and Kirchner, 2000; Neal *et al.*, 2007). Monitoring has typically been on a weekly to fortnightly basis.

The headwaters of the Severn comprise two main subcatchments, the Afon Hafren and the Afon Hore (areas 3.6 and 3.2 km², respectively), and one smaller one, the Nant Tanllwyth (0.9 km²). Details of the sites and durations of the monitoring programme are in Neal and Kirchner, (2000) and in Neal *et al.*, (2007), so only the salient features need be described here.

The upper Severn has relatively high rainfall of around 2500 mm year⁻¹ and runoff, typically about 2000 mm year⁻¹, is rapid due to the hard rock geology (Silurian and Ordovician slates and shales) and thin acidic and acid sensitive soils (peat, podzols and gleys). The upper part of the catchment comprises the Plynlimon massif of acid moorland and hag peat, while in the lower half is acid moorland on which a Sitka spruce forest was planted up to

60 years ago. The forestry practice follows the cycle of a 'working' plantation, and many parts of the forest have been cropped and replanted. The cropping of the forest has been gradual apart from that on the lower Hore where an experimental clear-fell occurred between 1985 and 1989.

Rain-gauges were sited one at the top and one at the bottom of each of the two catchments; also a cloud/mist collector was near the top and five primary stream water monitoring points were also distributed across the catchments. There were two stream monitoring points on the Afon Hafren, one at an upper site half way down the catchment where the moorland meets the forest and one at a lower site that includes the whole catchment. A further two sites were on the Afon Hore, one at an upper site half way down the catchment with a 50–50% mix of moorland and forest cover and one at a lower site that monitors the whole catchment. The fifth stream monitoring site is on the forested Tanllwyth near its junction with the Afon Hore. Also, a further series of five small (less than 5 ha) catchment sites, including two pairs with fell (f) and control (c) streams, examines the effects of felling on podzolic and gley soils. The paired sites on podzol are SE1f and SE3c, while the gleys sites are TanNc and TanSf. A further small catchment has been monitored within the Hore catchment where felling has been undertaken (south2Hore with an area of 54 ha). In addition, a series of boreholes has been established across the catchment and at the paired stream sites; these have been sampled for shallow and in some cases deep groundwater.

Iodine analysis

After 0.45 µm cellulose nitrate membrane filtering of the water samples, the filtrates were stored cold (4°C) in the dark prior to analysis. The river and groundwater samples were filtered in the field, while the rainfall, cloud, throughfall and stemflow samples were filtered in the laboratory, soon after return from the field. Sample storage was in chromic acid-cleaned glass bottles with glass stoppers.

Iodine analysis used an automated colourimetric (autoanalyzer 1) method based on the catalytic effect of iodide on the reaction between cerium IV and arsenic III (Truesdale and Smith, 1975; Jones *et al.*, 1982). Duplicate analyses were undertaken with calibration in the range 0 to 5 µg l⁻¹ and a detection limit of 0.1 µg l⁻¹. Intermediate cross-check standards were included every five samples in an analytical run and the analysis was repeated if the response for the duplicate samples differed visually from each other on the chart, or if peak shape was irregular in any way.

Previously, researchers have reported data using the present methodology as 'total iodine'. However, because the waters are filtered, it corresponds to dissolved iodine

and, indeed, may include reactive material within the colloidal fraction that can pass through standard 0.45 μm filters. This 'dissolved total iodine' measurement has been largely synonymous with the dissolved inorganic species (iodide plus iodate). However, as Jones *et al.* (2004) note, the method may also respond to organo-iodine compounds in the water, but the extent of the response varies substantially according to which particular organo-iodine compound is present in the water. Because of this, the 'dissolved total iodine' measure is really a 'total dissolved labile iodine' fraction. In this paper, the word 'iodine' is used to describe this total dissolved labile fraction, to avoid the text becoming cumbersome.

Results

GENERAL FINDINGS

The iodine concentrations in the waters sampled were up to 66 $\mu\text{g l}^{-1}$ and usually much lower (Table 1). The highest concentrations were observed for the atmospheric inputs, with data skewed to lower concentrations while the high concentrations occur at low volumes of catch (Fig. 1). Of the atmospheric sources, average iodine concentration in

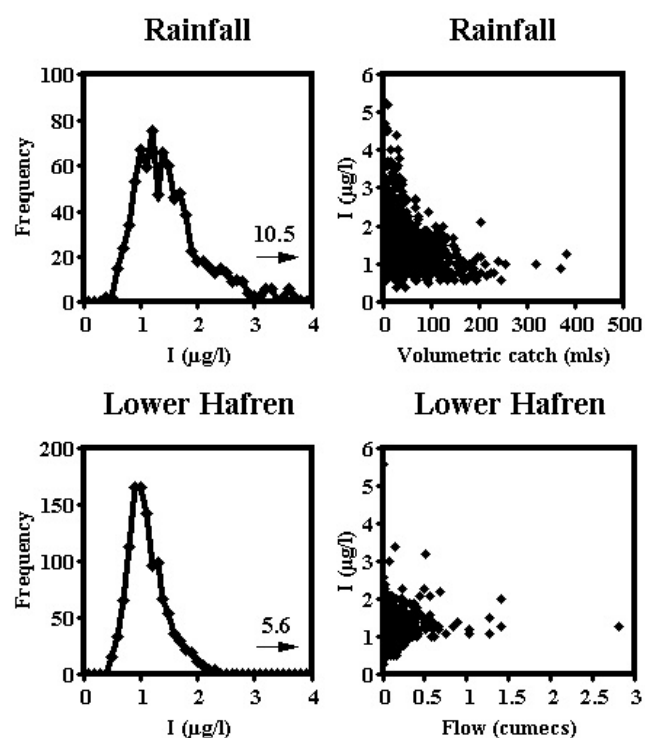


Fig. 1. The distribution of iodine concentration for rainfall and the lower Hafren with respect to frequency and volume of catch/flow.

Table 1. Iodine and DOC in Plynlimon waters.

	Average I	Average I, flow weighted	Minimum I	Maximum I	DOC	I/DOC
	$\mu\text{g l}^{-1}$	$\mu\text{g l}^{-1}$	$\mu\text{g l}^{-1}$	$\mu\text{g l}^{-1}$	mg l^{-1}	$\mu\text{g mg}^{-1}$
Rainfall	1.55	1.30	0.4	10.5	0.68	2.29
Cloud	10.55	8.43	0.8	66.0	2.42	4.36
Throughfall	5.46	4.40	1.2	33.3	11.40	0.48
Stemflow	8.16	6.40	1.4	42.0	19.15	0.43
Upper Hafren	1.00	1.08	0.5	4.5	1.42	0.71
Lower Hafren	1.15	1.29	0.4	5.6	1.64	0.70
Upper Hore	1.08	1.14	0.4	3.4	1.56	0.69
Lower Hore	1.07	1.18	0.3	3.2	1.51	0.71
Tanllwyth	1.89	1.83	0.9	4.8	3.25	0.58
South2Hore	1.34	1.30	0.7	6.4	1.74	0.77
SE1f	1.28	1.26	0.6	3.8	1.16	1.10
SE3c	1.05	1.20	0.7	2.5	0.62	1.71
TanNc	3.49	2.62	0.9	7.2	15.24	0.23
TanSf	4.22	3.34	0.9	14.1	12.78	0.33
HA4B	0.65	NA	0.4	1.4	0.63	1.02
Quarry	0.57	NA	0.3	1.3	0.42	1.34
SE1Bf	0.74	NA	0.4	2.3	0.59	1.25
SE3Bc	1.63	NA	0.8	3.0	0.42	3.87
TanNBf	1.10	NA	0.8	1.7	0.50	2.20
TanSBc	8.51	NA	0.6	21.0	1.48	5.74

rainfall is relatively low ($1.55 \mu\text{g l}^{-1}$) with cloud water, throughfall and stemflow (10.55 , 5.46 and $8.16 \mu\text{g l}^{-1}$, respectively) being three to eight times higher. Correspondingly, the stream waters have average concentrations lower than that for rainfall (typically 1.00 to $1.34 \mu\text{g l}^{-1}$) although, in some cases, average concentrations can be higher (1.89 to $4.22 \mu\text{g l}^{-1}$). The data for the streams are skewed but the skew is less than that for the atmospheric inputs (Fig 1). Also, the distribution of iodine in the streams is not linked systematically to flow. Rather, there is a particularly high scatter in iodine concentration to both high and low concentrations at low flows with concentrations levelling off to a value close to the average at high flows (Fig 1). For groundwaters, average concentrations are generally lower than those for the streams (0.57 to $1.10 \mu\text{g l}^{-1}$) but two sites are outliers with higher average concentrations (1.63 and $8.51 \mu\text{g l}^{-1}$). For both the streams

and the groundwaters, the most pronounced outliers are for the small forested catchment systems on gley soils. For the groundwaters, the frequency distribution for iodine concentrations is similar to that for the streams and there is no clear link between the distribution of iodine concentration and groundwater level.

The stream waters show a marked seasonal pattern. Iodine concentrations are higher in autumn (Figs. 2 and 3) and, for the long-term monitoring sites on the river, the iodine concentration increases with time in the upper and lower Hore and the lower Hafren and the Tanllwyth that drain from the forested areas. However, there is no clear trend for the upper Hafren site that drains the moorland area only. Statistical analysis reveals linear gradients for iodine over time averaging around $0.03 \mu\text{g l}^{-1} \text{ yr}^{-1}$ with a range of 0.025 ± 0.003 to $0.044 \pm 0.010 \mu\text{g l}^{-1} \text{ yr}^{-1}$ for the forested areas and a gradient of $0.008 \pm 0.005 \mu\text{g l}^{-1} \text{ yr}^{-1}$ for the moorland (Table 2).

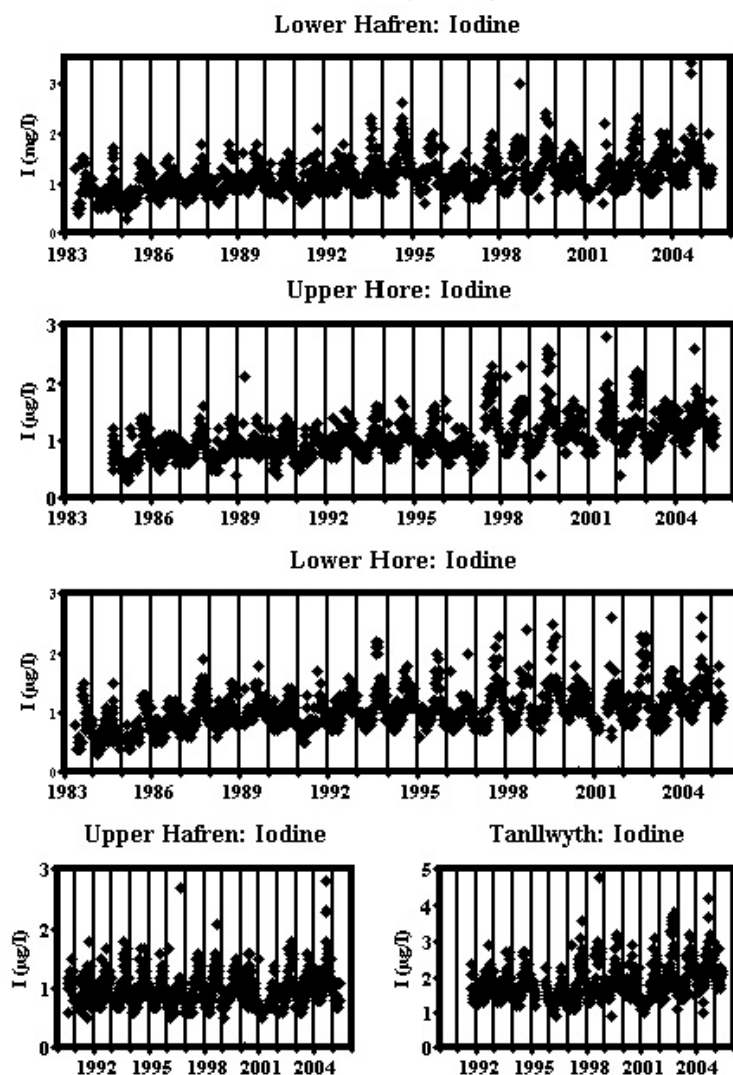


Fig. 2. Time series for iodine concentration for the main long-term stream sites in the upper Severn.

Table 2. Iodine variation over time for the main Plynlimon streams: linear regression relationships. The units are $\mu\text{g l}^{-1}$ for iodine and years for time.

	Study period	Gradient	2*std	Constant	2*std	r^2	N
Rainfall	1983 to 2005	-0.011	0.008	23.3	1.4	0.009	835
Cloud	1990 to 2005	-0.127	0.176	264.4	14.2	0.004	521
Upper Hafren	1990 to 2005	0.008	0.005	-14.0	0.6	0.011	738
Lower Hafren	1983 to 2005	0.025	0.003	-49.3	0.7	0.179	1168
Upper Hore	1984 to 2005	0.030	0.003	-57.9	0.6	0.251	1062
Lower Hore	1983 to 2005	0.025	0.003	-49.6	0.6	0.223	1168
Tanllwyth	1991 to 2005	0.044	0.010	-86.6	1.0	0.113	653

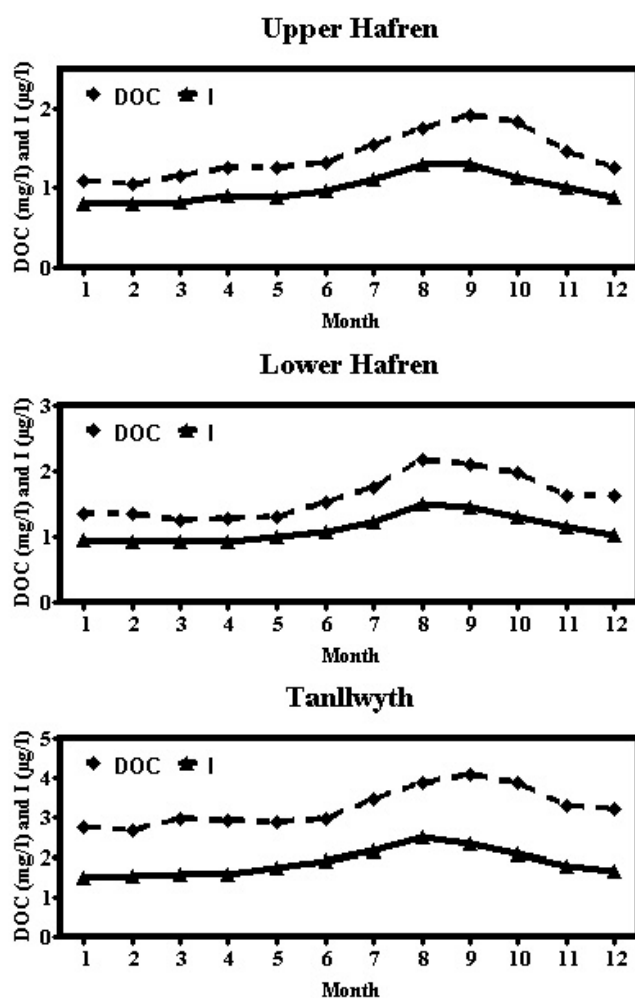


Fig. 3. The average monthly concentrations of iodine and DOC in the upper and lower Hafren and the Tanllwyth.

IODINE CONCENTRATION RELATIONSHIPS WITH CHLORIDE AND DOC

Rainfall, cloud water, throughfall and stemflow

At Plynlimon, iodine concentrations in the atmospheric inputs and stream and groundwater outputs are mainly of marine origin, but there can be a micro-organic (pollutant) and humic (vegetative) component as well. To examine the linkage of iodine with these two sources, the iodine concentrations have been compared with those of chloride that represents the marine contribution and with DOC that provides a measure of the organic carbon constituents.

For rainfall and cloud water, there is a scattered relationship between DOC and chloride with iodine although there is some structure of a wedge-shaped pattern and a corner that passes through the origin (Fig 4a). There are two features to this. Firstly, there is a lower boundary line for chloride that probably reflects a largely marine component. Secondly, there is an upper boundary that represents a DOC component. The chloride and DOC concentrations are negatively related since chloride comes from Atlantic frontal systems that are relatively unpolluted, while the DOC comes from contaminant sources (Wilkinson *et al.*, 1997). Thus, the pattern is one of two-component mixing with a dilution (washout) component superimposed. For throughfall and stemflow, the pattern is similar to that for chloride, but for DOC there is a much stronger relationship with iodine and a near linear feature was observed (Fig 4b).

To separate the influence of the marine and the organic component on the iodine signal, a multiple linear regression was undertaken. As there is an approximately inverse relationship between iodine concentration and volume of catch, the regression used the logarithm of the volume of the catch as this gave a more linear relationship. The results of this exercise show (Table 3):

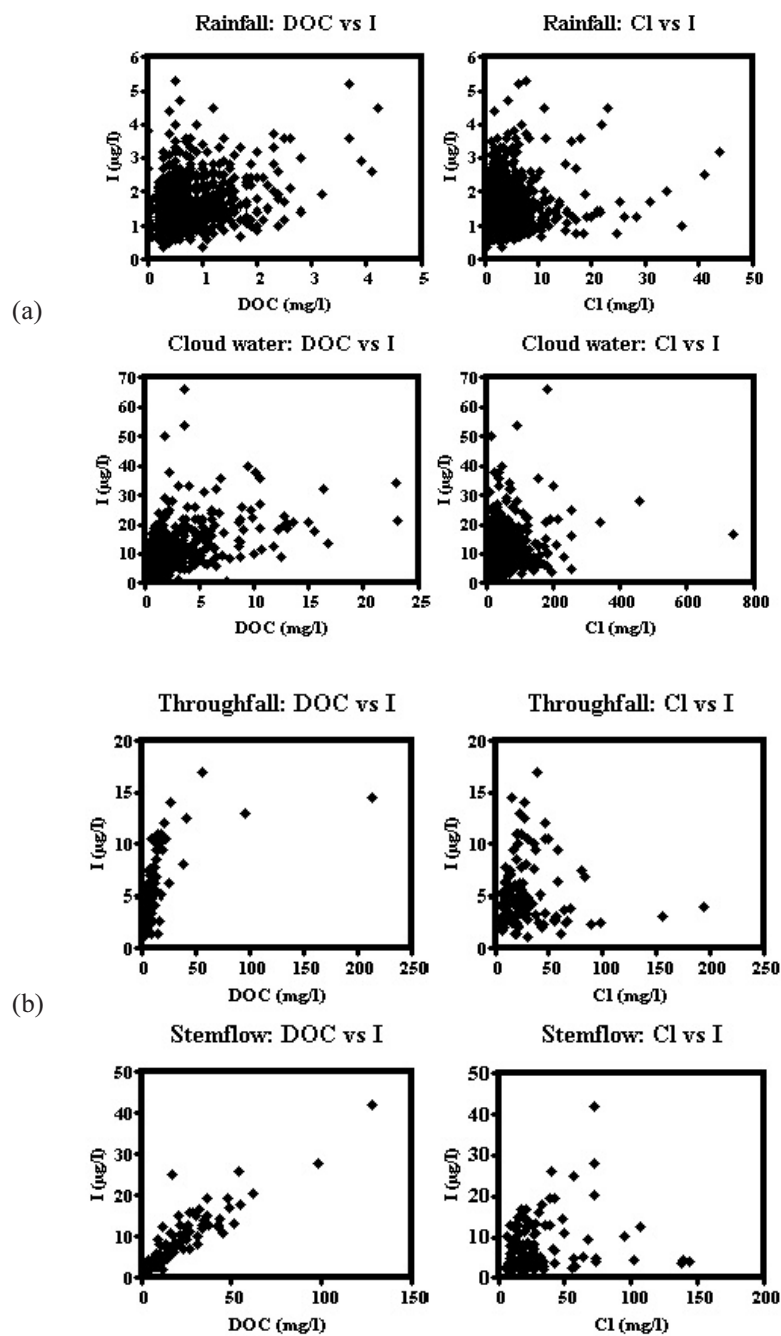


Fig. 4. DOC and chloride concentration relationships with iodine in (a) rainfall and cloud water, (b) throughfall and stemflow.

Table 3. Multiple linear regression analysis for the logarithm of the volume of catch and dissolved organic carbon and chloride concentration with total iodine. The units are $\mu\text{g l}^{-1}$ for iodine, mg l^{-1} for DOC and chloride and ml for volume of catch.

	<i>log(catch)</i>		<i>DOC</i>		<i>Cl</i>		<i>Constant</i>		<i>r</i> ²	<i>N</i>
	<i>Grad</i>	<i>2*std</i>	<i>Grad</i>	<i>2*std</i>	<i>Grad</i>	<i>2*std</i>	<i>Grad</i>	<i>2*std</i>		
Rainfall	-0.97	0.13	0.13	0.09	0.033	0.010	2.92	1.25	0.329	827
Cloud water	-1.08	0.79	1.14	0.19	0.028	0.009	9.72	12.00	0.286	583
Throughfall	-2.79	1.45	0.07	0.02	0.006	0.017	9.17	4.96	0.405	118
Stemflow	-1.82	2.17	0.28	0.03	0.027	0.021	5.15	6.15	0.779	112

- The multiple regressions are all significant statistically ($p < 0.001$).
- For rainfall and cloud water, all three regression terms are statistically significant, and the gradients are similar except for DOC where the rainfall gradient ($0.13 \pm 0.09 \mu\text{g}:\text{mg}$) is only about a ninth of that for the cloud water ($1.14 \pm 0.19 \mu\text{g}:\text{mg}$): for chloride, the respective gradients were 0.033 ± 0.010 and $0.028 \pm 0.009 \mu\text{g}:\text{mg}$.
- For throughfall and stemflow, the chloride coefficient is statistically much weaker than that for rainfall and cloud water and, in the case of stemflow, so too is the logarithm of the volume of catch. The clearest correlation is between DOC and iodine although the gradients differ. For throughfall, the gradient $\delta\text{I}/\delta\text{DOC}$ is $0.07 \pm 0.02 \mu\text{g}:\text{mg}$ and for stemflow it is $0.28 \pm 0.03 \mu\text{g}:\text{mg}$, the values straddling that for rainfall.

DOC in the atmospheric inputs is linked to several pollutants. To test if the iodine trend relates specifically to DOC, another pollutant component, non-marine sulphate was included in the multiple regression analysis. This exercise showed that for rainfall and cloud water, the non-marine sulphate term was statistically significant (gradient 0.10 ± 0.03 and $0.31 \pm 0.07 \mu\text{g}:\text{mg}:\text{SO}_4$, respectively) at the expense of the DOC (gradient 0.08 ± 0.09 and $-0.20 \pm 0.33 \mu\text{g}:\text{mg}$, respectively). The rainfall and cloud water gradients for chloride remain similar to those for the regression excluding the non-marine sulphate term (0.027 ± 0.009 and $0.031 \pm 0.008 \mu\text{g}:\text{mg}$, respectively). Correspondingly, for stemflow, non-marine sulphate correlates significantly with iodine within the regression equation (gradient $0.06 \pm 0.02 \mu\text{g}:\text{mg}$), but barely for throughfall (gradient $0.02 \pm 0.02 \mu\text{g}:\text{mg}$). However, for both throughfall and stemflow, the relationship between iodine and DOC is maintained and the gradient is very similar to that for the regression excluding the non marine sulphate term (0.07 ± 0.02 and $0.024 \pm 0.04 \mu\text{g}:\text{mg}$, respectively).

STREAM WATERS

Statistical analysis along the lines of that for rainfall, cloud water, throughfall and stemflow was undertaken for the stream waters. The analysis has been divided into two parts. Firstly, for the long term monitoring sites, multiple regression analysis has been undertaken for iodine with DOC, chloride and time. Time was included as the iodine concentration for most of these stream sites was increasing throughout the period of study (Fig. 2, Table 2). Secondly, multiple regressions have been undertaken for just DOC and chloride for all the monitoring sites: in the case of the shorter term sites, the longer term trends are not clearly marked – the longer-term sites are included for comparative purposes.

Long term stream monitoring sites

In all cases, the multiple regression analysis gives results that are highly significant statistically; iodine concentrations are correlated positively with DOC and negatively with chloride. Of the two, the $\delta\text{I}/\delta\text{DOC}$ term is statistically most significant (Table 4). The average $\delta\text{I}/\delta\text{DOC}$ gradient is around $0.185 \mu\text{g}:\text{mg}$ with a range of 0.120 ± 0.017 to $0.194 \pm 0.020 \mu\text{g}:\text{mg}$. For chloride, the average gradient is $-0.04 \mu\text{g}:\text{mg}$ with a range of -0.036 ± 0.022 to $-0.056 \pm 0.021 \mu\text{g}:\text{mg}$. Iodine concentrations are positively correlated with time for the catchments draining forest (lower Hafren, upper and lower Hore and the Tanllwyth), but not for the catchment draining moorland (upper Hafren). For the forested areas, the increase in iodine over time is around $0.015 \mu\text{g l}^{-1} \text{yr}^{-1}$ with a range of 0.013 ± 0.003 to $0.023 \pm 0.009 \mu\text{g l}^{-1} \text{yr}^{-1}$; the corresponding gradient for the moorland is $0.001 \pm 0.005 \mu\text{g l}^{-1} \text{yr}^{-1}$. These values are similar to about a half of those obtained by linear regression involving time only as the independent variable. Clearly the trend to increasing iodine concentrations over time is partly related to that for DOC increase.

Table 4. Multiple linear regression relationship between time (years), dissolved organic carbon and chloride (mg l^{-1}) with total iodine ($\mu\text{g l}^{-1}$) in Plynlimon streams for the long term monitoring sites.

	Year		DOC		Chloride		Constant		r^2	N
	Grad	2*std	Grad	2*std	Grad	2*std	Const	2*std		
U Hafren	0.001	0.005	0.173	0.020	-0.036	0.022	-1.4	0.3	0.323	738
U Hore	0.021	0.003	0.175	0.019	-0.048	0.010	-40.7	0.3	0.490	1057
L Hafren	0.013	0.003	0.185	0.016	-0.041	0.013	-25.3	0.3	0.471	1155
L Hore	0.016	0.002	0.194	0.020	-0.042	0.010	-30.2	0.3	0.480	1179
Tanllwyth	0.023	0.009	0.120	0.017	-0.056	0.021	-44.2	0.4	0.370	652

Table 5. Multiple linear regression relationship between dissolved organic carbon and chloride (mg l^{-1}) with total iodine ($\mu\text{g l}^{-1}$) in Plynlimon streams.

	Year		DOC		Chloride		Constant		r^2	N
	Grad	2*std	Grad	2*std	Grad	2*std	Const	2*std		
U Hafren	0.001	0.005	0.173	0.020	-0.036	0.022	-1.4	0.3	0.323	738
U Hore	0.021	0.003	0.175	0.019	-0.048	0.010	-40.7	0.3	0.490	1057
L Hafren	0.013	0.003	0.185	0.016	-0.041	0.013	-25.3	0.3	0.471	1155
L Hore	0.016	0.002	0.194	0.020	-0.042	0.010	-30.2	0.3	0.480	1179
Tanllwyth	0.023	0.009	0.120	0.017	-0.056	0.021	-44.2	0.4	0.370	652
			DOC		Chloride		Constant		r^2	N
			Grad	2*std	Grad	2*std	Const	Grad		
U Hafren			0.174	0.020	-0.036	0.022	0.96	0.26	0.323	738
U Hore			0.231	0.019	-0.033	0.011	0.95	0.28	0.387	1057
L Hafren			0.212	0.015	-0.040	0.014	1.08	0.29	0.428	1155
L Hore			0.243	0.019	-0.038	0.011	1.00	0.27	0.408	1179
Tanllwyth			0.132	0.017	-0.059	0.022	1.92	0.43	0.342	652
S2Hore			0.082	0.027	-0.049	0.008	1.53	0.29	0.287	637
SE1f			0.287	0.062	0.032	0.024	0.68	0.32	0.348	174
SE3c			0.237	0.091	0.031	0.029	0.67	0.21	0.193	124
TanNc			0.145	0.041	-0.090	0.067	2.01	1.19	0.496	106
TanSf			0.201	0.065	-0.271	0.144	3.45	1.74	0.526	133

Long and short term stream monitoring sites

For all the streams, the multiple regression analysis is significant statistically and the greatest component of that significance is DOC (Table 5). The gradient $\delta\text{I}/\delta\text{DOC}$ averages around $0.20 \mu\text{g}:\text{mg}$ with a range of 0.082 ± 0.027 to $0.287 \pm 0.062 \mu\text{g}:\text{mg}$. Correspondingly, for chloride, there is a negative gradient between iodine and chloride in all but two cases. Where the relationship was negative, the average gradient $\delta\text{I}/\delta\text{Cl}$ was typically around $-0.045 \text{ lg}:\text{mg}$ with a range of -0.036 ± 0.022 to $-0.271 \pm 0.144 \mu\text{g}:\text{mg}$; the most negative value was atypical as DOC concentrations were especially high. For the two small streams draining podzolic soils (SE1 and SE3), iodine was related positively with gradient $\delta\text{I}/\delta\text{Cl}$ around $0.032 \pm 0.029 \mu\text{g}:\text{mg}$ but barely significant statistically.

For the long term sites, the exclusion of time from the regression analysis led to slightly higher but variable $\delta\text{I}/\delta\text{DOC}$ gradients (0 to 25%), but the $\delta\text{I}/\delta\text{Cl}$ gradient was largely unaffected.

Seasonal patterns between iodine and DOC

Both the iodine and the DOC concentrations change similarly through the year, with the highest concentrations occurring between July and November (Fig. 3). Superimposed on this pattern is a hysteresis effect (Fig. 5), where during the period May to August, iodine concentrations are relatively high for a given DOC

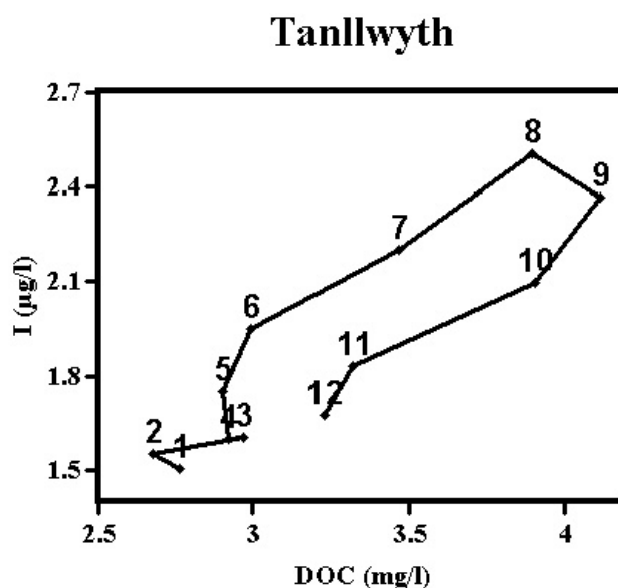


Fig. 5. The relationship between DOC and iodine for monthly averages for the lower Hafren. The data points are joined by lines for consecutive months through the years and the data point for each month is labelled by the month number.

concentration. This feature indicates that during the growing season either iodine is enriched relative to DOC or there is a net loss of DOC relative to iodine.

GROUNDWATER

There is no clear statistically significant relationship for either DOC or chloride with iodine concentration in groundwater and no clear relationship with groundwater depth.

Mass balance input output relationships

Previous studies at Plynlimon based on the early record indicate a net uptake of iodine by the catchment (Neal *et al.*, 1990; Durand *et al.*, 1994). Here, the analysis is updated using information on flow-weighted averaged iodine concentrations for rainfall, cloud water and stream water (Table 1) coupled with hydrological information on rainfall and cloud deposition and stream runoff (Neal *et al.*, 2007). For both the upper and lower Hafren and the Hore, there is a net uptake of about 40% of the input. Correspondingly, the net uptake for the main stem of the Tanllwyth is around 22% and for the small Tanllwyth streams, TanNc and TanSf, there is a net release of 11 and 42%, respectively. The lower uptake for the main stem of the Tanllwyth and the net release of iodine from the catchment for the small Tanllwyth streams results from the higher flow-weighted iodine concentrations in the stream and is associated with a high DOC content in the waters (Table 1). For the Tanllwyth, with its gley soils and reducing conditions, both DOC and iodine would be expected to be mobilised.

Discussion

The average iodine concentration in rainfall at Plynlimon, $\sim 1.55 \mu\text{g l}^{-1}$ is similar to that observed for the English Lake District (Jones and Truesdale, 1984). It is also similar to other parts of the UK where iodine concentrations average 1.5 to $2.5 \mu\text{g l}^{-1}$ in coastal areas and $1 \mu\text{g l}^{-1}$ inland (Whitehead, 1984; Fuge, 1989; Edmunds and Smedley, 1996). The iodine in rainfall is associated with both marine and anthropogenic sources and the inverse relationship between iodine concentration and volume of catch represents a 'wash out' of mist and particulate components during rainfall events. For cloud water, throughfall and stemflow, iodine concentrations are enriched relative to rainfall; again, there is both a marine and an anthropogenic component. However, in the cases of throughfall and stemflow, there is a very strong linkage between iodine and DOC concentrations; it may well be that much of the iodine is associated with organic material cycled by the trees and washed from their surfaces. This organic carbon is brown in colour and linked to humic acids. For rainfall, cloud water,

throughfall and stemflow, even in the cases where iodine concentration is low and chloride concentration is high, the iodine to chloride ratio is still over an order of magnitude higher than that in seawater. For example, in rainfall, the ratio for the lower iodine to chloride bound is $0.04 \mu\text{g:mg}$ and that for the regression gradient is $0.033 \pm 0.010 \mu\text{g:mg}$, which compares with a ratio for sea water of $0.0026 \mu\text{g:mg}$. This illustrates that, for the atmospheric input, even the marine component is enriched in iodine relative to seawater.

Stream water iodine concentrations averaged around $1.5 \mu\text{g l}^{-1}$; this value is similar to that in the English Lake District (Jones and Truesdale, 1984) but is perhaps on the low side for river systems in general. For example, Moran *et al.* (2002) reported that while iodine concentration across the rivers of the world was mostly $< 20 \mu\text{g l}^{-1}$, the median and mean were 10.2 and $19.9 \mu\text{g l}^{-1}$, respectively. Stream water iodine concentrations have been increasing over time for the Plynlimon areas draining forest (with or without harvesting) and this increase has been observed for bromine, DOC and iron as well (Neal *et al.*, 2005). With felling and tree aging, transpiration decreases (Marc and Robinson, 2007). The increases in iodine, bromine, DOC and iron over time are consistent with the effects of the forest soils wetting up and releasing these components. This feature also explains why the moorland system does not show similar hydrochemical changes. As to the extent of the felling effects, detailed analysis of the Hore clear-fell indicated previously that iodine increases from 0.8 to $1.2 \mu\text{g l}^{-1}$ within the forest streams (Neal *et al.*, 1992). However, most of the signal seems to lie within the noise of the background trend and even with the paired catchment studies there is little discernable change. The original analysis of the felling effect for the Hore was undertaken for pre-1992 data; here, the data were reanalysed with more than ten years of additional information. To do this, the time series for the lower Hafren and the lower Hore were compared (the lower Hafren can be considered a control site for the lower Hore). Regression analysis revealed a strong linear relationship with no clear-felling effect. The relationship was

$$I_{\text{Hore}} = 0.80 \pm 0.03 * I_{\text{Hafren}} + 0.15 \pm 0.37, r^2 = 0.717, N = 1154.$$

Comparing annual average concentrations, between 1983 and 1990 (when the Hore was clearfelled), the iodine concentration changed approximately linearly from around -17% to around -6% ($100 * (\text{Hore-Hafren}) / \text{Hafren}$). After that time, there was no systematic change with the differences fluctuating between -10 and $+5\%$. Thus, in the light of the longer run of data, even for the lower Hore with the control clear-fell, the changes involved were small.

The negative relationship between iodine and chloride

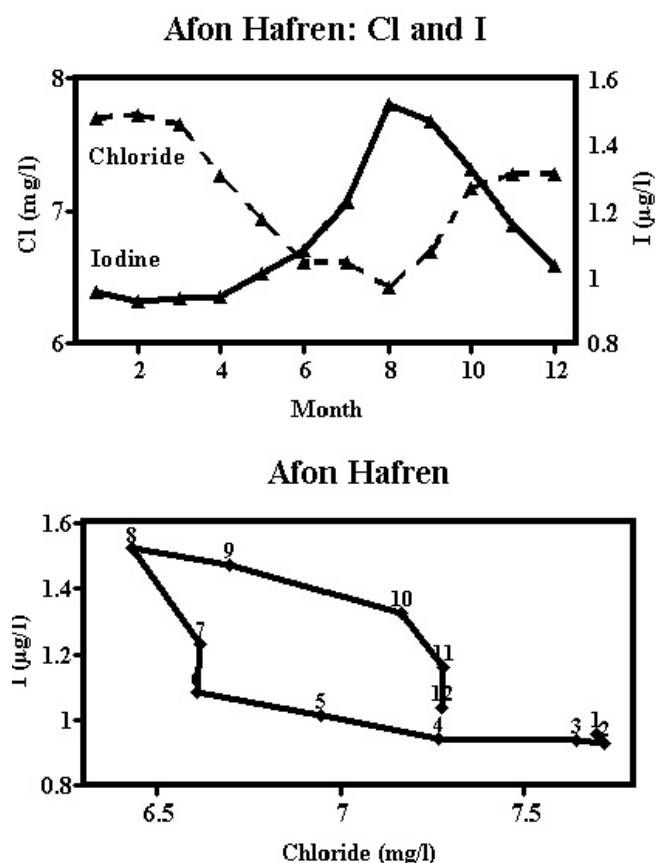


Fig. 6. Variations in average chloride and iodine concentration with season for the lower Afon Hafren. The upper plot shows the variation of chloride and iodine with month. The lower plot is of chloride versus iodine: the data points are joined in monthly sequence with the months numbered.

concentrations in the stream reflects the occurrence of Atlantic frontal systems enriched in sea salts that occur in winter when iodine concentrations are at their lowest. Figure 6, for the lower Afon Hafren, shows that monthly average chloride concentrations are lower in summer between April and October, while iodine concentrations peak from July to November. Correspondingly, a plot of iodine against chloride concentration for monthly averages shows a strong clockwise hysteresis.

Within the groundwater areas, DOC decomposes and the waters become high in dissolved carbon dioxide (Neal *et al.*, 1997). Because of this, the correlations of iodine with DOC and chloride are largely lost. The lower iodine concentrations in the groundwaters relative to the stream imply a loss of iodine (as well as DOC) in the groundwater areas. At only one site was the iodine concentration in groundwater relatively high and, at this site, DOC was also relatively high. This situation occurred for the gley area and it probably reflects unusually wet and highly reducing conditions with increased iodine mobilisation. Even

including this outlier value, the concentrations of iodine in the groundwaters at Plynlimon are relatively low, and much lower than for many UK aquifers. For example, high iodine concentrations (median concentration $32 \mu\text{g l}^{-1}$) occur within the Chalk, probably due to oxidation of organic matter, while for non-carbonate aquifers, iodine concentrations are typically a quarter that for the Chalk (Edmunds *et al.*, 1989; Edmunds and Smedley, 1996).

Overall, the present study extends the findings of Neal *et al.* (1990) in respect of the iodine hydrochemistry of a major upland research site in mid-Wales. It illustrates the need for long term studies to extend the patterns of change beyond the seasonal and year-to-year level. A gradual increase in iodine concentration in the streams seems to be related to catchment wetness. The work also provides a clear indication of the importance of maritime and pollutant components of iodine in the rainfall and cloud water inputs to the Plynlimon catchments and, potentially, of chemical cycling of iodine through the vegetation. Iodine accumulates in organic-rich soils and it is known that, over the last 50 years worldwide, the ratio of $^{129}\text{I}/^{127}\text{I}$ has increased by an order of magnitude due to atmospheric atomic tests and point sources of pollution (e.g. Sellafield in the 1950s). Thus processes that lead to increased wetting (e.g. the habitat restoration of previously drained peat-lands by flooding, or tree harvesting) should be questioned within the vicinity of known point sources of radiological pollution and perhaps even some diffuse ones.

The method used for iodine assay is an operational one and it cannot be concluded that the method determines only inorganic iodine within the stream. In a companion study, issues of organo-halogen components within the Plynlimon inputs and stream and groundwater outputs have been raised in respect of bromine (Neal *et al.*, 2007). There may well be a significant organic iodine component within the throughfall and stemflow monitored within the present study. For both iodine and bromine, the current Plynlimon studies provide an important baseline and key hydrochemical information. However, for further progress to be made, inorganic and organic speciation studies of the halogens are required.

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